

CATALOG DESCRIPTION

Part two of the study of carbon compound chemistry covering: structure and reaction mechanisms of carboxyl, amine, conjugated, and polyfunctional systems; ultraviolet spectroscopy; biochemistry; and synthetic polymers.

Prerequisites: CHEM 251 or equivalent

Semester Offered: Spring



Course Learning Outcomes

Upon successful completion of the course, the student will be able to...

- 1. Propose Williamson ether synthesis of ethers.
- 2. Show the mechanisms of acid catalyzed and base catalyzed ring opening of epoxides.
- 3. Show the mechanism of epoxidation and acid catalyzed ring opening and cyclization related to the biosynthesis of steroids.
- 4. Show the mechanism of the formation of epoxy ether based polymers.
- 5. Construct molecular orbitals and electronic configurations of conjugated systems.
- 6. Predict the products of Diels-Alder reactions.
- 7. Use HOMO-LUMO interactions to predict thermal or photochemical cycloadditions.
- 8. Predict UV absorption maxima of conjugated systems.
- 9. Construct and interpret molecular orbitals and electronic configurations of aromatics.

- 10. Use the polygon rule on conjugated cyclic systems to determine aromaticity.
- 11. Determine the aromaticity of heterocyclic compounds and nitrogen acidity.
- 12. Use IR, NMR, UV, and MS to determine the structures of aromatic compounds.
- 13. Predict products and give mechanisms for electrophillic aromatic substitutions.
- 14. Design syntheses that use the influences of substituents to generate the correct isomers of multisubstituted aromatic compounds.
- 15. Explain how Friedel-Crafts Acylation overcomes two of the three limitations of Friedel-Crafts Alkylation
- 16. Show how to synthesize ketones and aldehydes from oxidation of alcohols, ozonolysis, Friedel-Crafts acylations, organolithiums, and acid chlorides.
- 17. Show mechanisms of nucleophillic additions and condensation reactions.
- 18. Interpret the IR, NMR, UV, and MS of ketones and aldehydes.
- 19. Predict the products of McLafferty rearrangement in the Mass Spectrometer
- 20. Predict the approximate maxima for allowed and forbidden electronic transitions.
- 21. Show how to synthesize amines by reductive amination and acylation-reduction.
- 22. Predict the basicity of amines.
- 23. Use amines in synthesis.
- 24. Interpret the IR, NMR, UV, and MS of amines.
- 25. Name and identify acids, acid chlorides, anhydrides, amides, esters, carbonates, and urethanes.
- 26. Use carboxylic acids and derivatives in fisher esterification and hydrolysis reactions.
- 27. Show how to interconvert acid derivatives by Nucleophillic Acyl Substitution.
- 28. Interpret the IR, NMR, UV, and MS of carboxylic acids.
- 29. Identify and name essential features of carbohydrates and nucleic acids.
- 30. Determine the structures of the anomers and epimers of glucose.
- 31. Name monosaccharides and disaccharides, and draw their structures from their names.
- 32. Predict reaction products and write mechanisms involving carbohydrates.
- 33. Recognize the structures of DNA and RNA, and draw the structures of a ribonucleotide and deoxyribonucleotide.
- 34. Name amino acids and peptides, and draw the structures from their names
- 35. Explain which amino acids are acidic, basic, or neutral.
- 36. Show how an amino acid is synthesized.
- 37. Show how classical and solid-phase peptide synthesis would be used to make a given peptide.
- 38. Discuss and identify the four levels of protein structure.
- 39. Classify lipids.
- 40. Predict physical properties of fats and oils.
- 41. Identify isoprene units in terpenes.
- 42. Explain how soaps and detergents work.
- 43. Synthesize and analyze, ethers, aromatics, carbonyls, and amines in the laboratory.